TETRAHYDROFRANKLINONE

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THE essential oil of Acradenia franklinii (Kippist) contains a mixture of ketones based on a phloroglucinol nucleus. Our interest in one of these, franklinone, arose out of the close similarity of the proposed structure (I) to a degradation product of the hop β -acid, colupulone (II; $R = Pr^i$) studied earlier in these laboratories. Acid degradation of colupulone gave two pyrans (III; $R = Pr^i$) and (IV; $R = Pr^i$, R' = H); the former being converted into the latter by treatment with concentrated sulphuric acid. Methylation of the latter using diazomethane gave (IV; $R = Pr^i$, R' = Me) which had light absorption, λ_{max} 275 m μ (log \in 3.6), in agreement with that reported for tetrahydrofranklinone, allower homologue.

M.E. Baldwin, I.R.C. Bick, A.A. Komzak and J.R. Price, Tetrahedron 16, 206 (1961).

² G.A. Howard, J.R.A. Pollock and A.R. Tatchell, <u>J. Chem. Soc.</u>
174 (1955).

³ B.H. Arnold, J.J.H. Hastings and T.K. Walker, <u>Chem. & Ind</u>. 323 (1955).

We now report the synthesis of tetrahydrofranklinone by a route analogous to that used to prepare the lupulone degradation products. The reaction of phloroacetophenone with two moles of 1-bromo-3-methylbut-2-ene in dry chloroform afforded the pyran (III; R = Me), m.p. 117-8°, y max 1600 cm⁻¹ (chelated C=0), which was rearranged by treatment with concentrated sulphuric acid into the pyran (IV; R = Me, R' = H), m.p. 159-159.5°, y max 3400 cm⁻¹ (OH) 1650 cm⁻¹ (aryl C=0). Methylation of the latter compound using diazomethane gave tetrahydrofranklinone (IV; R = R' = Me), m.p. 141.5-142°, undepressed on admixture with authentic tetrahydrofranklinone; x max 275 m/m (log \in 3.6). The infra-red spectra of the authentic and the synthetic tetrahydrofranklinone were identical.

Acid degradation of synthetic 4 acetolupuphenone

⁴ W. Riedl, <u>Ber. 85</u>, 692 (1952).

(II; R = Me) afforded the same two pyrans, (III; R = Me) and (IV; R = Me, R' = H).

Attempts to prepare franklinone itself by bromination and subsequent dehydrobromination of tetrahydrofranklinone have so far been unsucessful. All the new compounds described analysed satisfactorily.

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